Improvement of biogas production from oil palm empty fruit bunches (OPEFB)

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ABSTRACT

Oil palm empty fruit bunches (OPEFB), a waste lignocellulosic material, which is the main byproduct of vegetable oil production industries in Indonesia and Malaysia, was utilized as a source for biogas production. Pretreatments using NaOH as well as phosphoric acid were investigated to improve the biogas production. Clear positive effects of the pretreatments on the yield of methane were observed. The best improvement was achieved when 8% NaOH for 60 min was used for the pretreatment, which resulted in 100% improvement in the yield of methane production. In addition, treatment with phosphoric acid resulted in 40% improvement in the methane yield compared with that of the untreated material. The results showed that the carbohydrate content of OPEFB could be efficiently converted to methane under the anaerobic digestion process. 97% of the theoretical value of methane production was achieved after the pretreatment with NaOH for 60 min. Moreover, the initial rate of methane production was also increased by more than 85% after the treatment with NaOH compared with that of the untreated OPEFB.

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1. Introduction

The major crop plantations in Indonesia are the oil palm (Elaeis guineensis) cultivations spread along the islands of Sumatra and Kalimantan. Its valuable oil producing fruits are the main source of vegetable oil production in the world. Palm oil production in Indonesia and Malaysia dominates the vegetable oil market, counting up to 35 million tons of crude palm oil (CPO) production in 2010 (USDA, 2010). During the past years a significant change in the palm oil industry has taken place as CPO production in Indonesia exceeded the production in Malaysia and now Indonesia is the world leader (USDA, 2007).

Oil palm empty fruit bunch (OPEFB) is one of the main waste products in the oil palm industry and at present it is mainly incinerated causing air pollution problems around the palm oil mill areas (Rahman et al., 2006). OPEFB is a non-wood lignocellulosic material, which can be utilized as a renewable energy source for biogas production.

Biogas is commercially produced from energy crops and waste materials such as manure, wastewater sludge, and municipal solid waste. However, the biodegradability of different waste streams differs depending on their composition. Sugars and starch are easy degradable, lipids and proteins are intermediately degraded, while cellulose is not easily degradable during anaerobic conditions.

Buffiere et al. (2006) found a direct correlation between the absolute biodegradability and the sum of the lignin and cellulose content of different waste streams. The higher the lignin and cellulose content was the lower biodegradability was obtained. Therefore, in order to improve the biodegradability of lignocellulosic materials a pretreatment is needed to open up their compact structure (Taherzadeh and Karimi, 2008). There are several methods has been investigated for the pretreatment (Taherzadeh and Karimi, 2008; Xiao and Clarkson, 1997) including mechanical, physical, thermal, and chemical (i.e., alkali, acidic, oxidative) as well as biological methods. These different pretreatment methods are well investigated for ethanol processes, however, only a few studies exist regarding the enhancement of biogas production.

Mechanical pretreatments result in no inhibitors and by reducing the particle size of the substrates usually lead to increased methane production. However, these methods have high energy demands and are not economically attractive (Cowling and Kirk, 1976; Hendriks and Zeeman, 2009).

Among the chemical pretreatment methods, treatments with alkaline have been proved to effectively improve the biological conversion of lignocelluloses. The method was shown to be a promising pretreatment for improvement of anaerobic digestion of newspaper, corn stalk, hardwoods, softwood, and paper tubes (Fox et al., 2003; Qingming et al., 2005; Mirahmadi et al., 2010; Teghammar et al., 2009). Pretreatments with NaOH can be classified into “high concentration” and “low concentration” processes (Mirahmadi et al., 2010). Low-concentration NaOH pretreatment (typically with NaOH of 0.5–4%) needs a high temperature and...
pressure to be efficient, and no NaOH reuse is possible, since the mechanism is a reactive destruction of lignocelluloses. The process can efficiently disintegrate the hemicellulose and lignin (Fengel and Wegener, 1984; Taherzadeh and Karimi, 2008). However, different inhibitory components can be produced during the process. On the other hand, high-concentration NaOH pretreatment (typically with NaOH of 6–20%) at ambient pressure and relatively low temperatures can dissolve the cellulose and regenerate it. This process is very effective for the reduction of cellulose crystallinity leading to improvement in biological conversion of lignocelluloses (Mirahmadi et al., 2010). One of the advantages of high concentration NaOH pretreatment is the possibility of reusing the NaOH solution, which is very important regarding the economy and environmental impact of the process (Mirahmadi et al., 2010). As far as our knowledge, the high concentration NaOH pretreatment was not truly studied for improvement of biogas from lignocellulosic agricultural residues.

Phosphoric acid is among the most efficient pretreatment processes, which has been studied for improvement of enzymatic hydrolysis of lignocellulosic materials (Zhang et al., 2007). The process is able to disrupt the lignocelluloses’ structure and eliminate the resistance of lignin and hemicelluloses. One of the main advantages of this method is the possibility of recycling and reusing the phosphoric acid (Zhang et al., 2007).

The current work deals with production of biogas from OPEFB. Two different pretreatment, alkali with NaOH and phosphoric acid processes, were used for the improvement of the anaerobic digestion. Moreover, the effects of treatment retention times in the alkali pretreatment on the yield of methane accumulation were studied. Furthermore, changes in the composition and structure of the waste as results of the treatments were also investigated.

2. Materials and methods

2.1. Raw material

The OPEFB was obtained from a palm oil mill in Sumatra, Indonesia, and dried at 44°C. The feedstock was ground to pass a 40 mesh screen, which resulted in particle size of less than 0.420 mm. The material was then stored in an airtight plastic bottle at room temperature before use.

2.2. Pretreatment procedures

2.2.1. Concentrated phosphoric acid pretreatment

The OPEFB was treated with concentrated phosphoric acid according to Zhang et al. (2007). 10 g of the OPEFB was mixed for 10 min with 80 mL of phosphoric acid (85.7%) using a glass rod. The mixture was then incubated in a shaker bath (Grant OLS200, Grant Instruments Ltd., UK), at 90 rpm and 50°C for 30 min. In order to precipitate the cellulose, 200 mL of pre-cooled acetone was added to the mixture followed by centrifugation at 3400 rpm and room temperature for 20 min. This procedure was repeated three times until a clear supernatant could be obtained. The solid pellets were collected and washed several times with distilled water followed by vacuum filtration until pH 7 in order to remove the remaining acetone and acids. The treated material was then kept at 4°C until use.

2.2.2. Sodium hydroxide pretreatment

5 g OPEFB and 95 g NaOH solution (8% w/v) was mixed for 10 min in room temperature. Then, the mixture was incubated at 100°C for different periods of time, i.e., 10, 20, 40, or 60 min and mixed in every 10 min during the incubation period. The mixtures were then centrifuged at 10,000 rpm and room temperature for 6 min and washed with distilled water followed by vacuum filtration until pH 7 had been reached. The solids were kept at 4°C until use.

2.3. Production of biogas

Anaerobic batch digestions of the treated vs untreated samples were carried out at thermophilic conditions (55°C) according to the method described previously by Hansen et al. (2004). The inoculum was obtained from a large scale biogas plant (Borås Energi och Miljö AB, Sweden), operating at thermophilic conditions. The batch digesters were 118 mL serum glass bottles, closed with butyl rubber seals and aluminum caps. Each flask contained 20 mL of inoculum to which a certain amount of substrate was added to keep a VS ratio of 2:1 between the inoculum and the substrate added followed by the addition of deionized water to a final volume of 25 mL. A blank with only 5 mL deionized water and 20 mL inoculum was also performed in order to determine the gas production of the inoculum by itself. Each sample set up was performed in triplicates. Anaerobic conditions were provided by flushing the headspace of the bottles with 80% nitrogen and 20% carbon dioxide gas mixture. The samples were then incubated at 55°C for 30 days. Gas samples were taken regularly and analyzed for methane and carbon dioxide content using gas chromatography.

2.4. Analytical methods

The treated and untreated OPEFB were analyzed for carbohydrates and lignin fractions according to Sluiter et al. (2008). Based on this method, the lignocellulosic material was first degraded using two-step acid hydrolysis. The formed sugars were then quantified by HPLC. Furthermore, the acid-soluble and acid-insoluble lignin contents of the samples were determined with UV spectrosocopy and after drying the samples at 575°C, respectively.

High performance liquid chromatography (HPLC, Alliance 2695, Waters, Milford, MA) was used to quantify sugars. It was equipped with a refractive index (RI) detector (Waters 2414) and an ion-exchange column (Aminex HPX-87P, Bio–Rad, USA), using pure water as eluent with a flow rate of 0.6 mL/min at 85°C.

Ash, total solids, and volatile solids content of the treated and untreated OPEFB were analyzed according to Sluiter et al. (2005).

The composition of gas produced in anaerobic digestion was analyzed by a gas chromatograph (Auto System Perkin Elmer, USA), which was equipped with a packed column (Perkin Elmer, 6’×1/8 OD, 80/100, Mesh, USA) and a thermal conductivity detector (Perkin Elmer, USA). Nitrogen with a flow rate of 23 mL/min at 60°C was used as carrier gas. The injections were carried out by using a pressure-tight carrier (VICI, Precision Sampling Inc., USA) making it possible to take gas samples at the actual pressure. The sample volume was 250 μL and the injector temperature was 150°C. Pure methane was used as standard gas and all the results of methane volumes are presented at standard conditions.

Fourier transform infrared (FTIR) spectrometer (Impact 410, Nicolet Instrument Corp., Madison, WI) was used for determining the changes in structure of OPEFB after the pretreatments. Each spectrum was obtained with an average of 60 scans and a resolution of 4 cm⁻¹, from 500 to 4000 cm⁻¹. Nicolet OMNIC 4.1 analyzing software was used for correcting baseline, smoothing of the spectra, and normalizing the spectra by the highest band of around 1050 cm⁻¹.

3. Results

3.1. Effects of pretreatment on OPEFB composition and structure

The composition of OPEFB samples before and after pretreatment was analyzed for lignin and carbohydrates content and the results are presented in Table 1.
Table 1
Total carbohydrate and lignin in the untreated and pretreated OPEFB expressed as percentage of TS.

<table>
<thead>
<tr>
<th>Material</th>
<th>Glucan</th>
<th>Xylan</th>
<th>Galactan</th>
<th>Mannan</th>
<th>AIL</th>
<th>ASL</th>
</tr>
</thead>
<tbody>
<tr>
<td>Untreated</td>
<td>32.4</td>
<td>6.8</td>
<td>1.0</td>
<td>0.9</td>
<td>23.0</td>
<td>1.6</td>
</tr>
<tr>
<td>H3PO4 pretreated</td>
<td>43.1</td>
<td>6.3</td>
<td>1.1</td>
<td>1.3</td>
<td>22.5</td>
<td>1.5</td>
</tr>
<tr>
<td>NaOH pretreated for:</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>10 min</td>
<td>43.0</td>
<td>7.6</td>
<td>0.4</td>
<td>2.2</td>
<td>22.1</td>
<td>1.4</td>
</tr>
<tr>
<td>20 min</td>
<td>47.7</td>
<td>7.6</td>
<td>0.7</td>
<td>2.3</td>
<td>22.2</td>
<td>1.3</td>
</tr>
<tr>
<td>40 min</td>
<td>50.0</td>
<td>5.2</td>
<td>0.0</td>
<td>1.0</td>
<td>21.4</td>
<td>1.1</td>
</tr>
<tr>
<td>60 min</td>
<td>53.0</td>
<td>4.8</td>
<td>0.0</td>
<td>1.4</td>
<td>20.7</td>
<td>1.0</td>
</tr>
</tbody>
</table>

* A Acid insoluble lignin.
* B Acid soluble lignin.

The untreated material contained 24.6% lignin and the dominant type of lignin in the waste was acid-insoluble. As expected, no significant change in the content of acid-insoluble lignin was measured after phosphoric acid pretreatment, while some lignin removals were observed in pretreatment with NaOH for longer times, i.e., 40 and 60 min.

Glucan, counting up to 32.4% of total solids (TS), was the major carbohydrate of the OPEFB. Other important carbohydrate components were xylan (6.8%), galactan (1.0%), and mannan (0.9%). All these sugars together were counting up to 41.1% carbohydrate content of TS as it is shown in Table 1. Unlike the content of lignin, the pretreatments affected the content of carbohydrates. Phosphoric acid pretreatment increased the glucan content to 43.1%, while it did not significantly change the composition of the other sugar polymers. The alkali pretreatment for 10 min increased all of the carbohydrates, except galactan. Prolongation of the alkali treatment for 40 or 60 min, resulted in higher glucan and mannan content and lower xylan and galactan content.

Total solid content and volatile solid (VS) in the untreated and pretreated OPEFB are presented in Table 2. The VS content of the untreated OPEFB was 84.1% of the TS content. The pretreatment with acid resulted in a reduced VS content counting up to 78.8% of the TS, while the alkali treatment increased the VS content of TS. The highest VS values observed were after pretreated with NaOH for shorter times, 10 and 20 min, when more part of the TS was organic material, since more than 98% VS content of TS was measured.

The structural changes in OPEFB were followed by FTIR spectroscopy. The spectra and the absorption data of untreated OPEFB and the one that treated with NaOH for 60 min are presented in Fig. 1 and Table 3, respectively. The absorption bands at 1430 and 898 cm⁻¹, which were assigned to the respective crystalline cellulose I and cellulose II, were used to study crystallinity changes (Colom et al., 2003). The absorbance ratio A1430/A898 or crystallinity indexes (CI) were 0.81 and 0.72 for untreated and the NaOH treated OPEFB, respectively, indicating that the treatment with NaOH resulted in reducing the absorption band at 1430 cm⁻¹ (cellulose I) and increasing the band at 898 cm⁻¹ (cellulose II).

Total crystallinity can also followed by the ratio of A1375/A2900 (Carrillo et al., 2004). This factor was also reduced from 1.045 to 0.462, when the material was treated with NaOH. There were also lower values observed for the bands characteristic for lignin and hemicellulose functional groups in the treated materials (Table 3).

3.2. Biogas production

All pretreated and untreated OPEFB samples were subjected to anaerobic digestion in batch reactors. Blanks and references were also included in the anaerobic batch digestion experimental series. The experiments were performed at thermophilic conditions during 30 days of cultivation period. The results of initial and final yields of methane are summarized in Fig. 2.

The initial reaction rate, calculated based on the accumulated methane production during the first 7 days, was increased by all the pretreatments. After 7 days of incubation, the yield of methane was 0.13Nm³/kg VS for untreated OPEFB, while minor improvement was observed with phosphoric acid pretreatment (0.15 Nm³/kg VS). The best results of methane production were achieved when the material was pretreated with NaOH (Fig. 2). Prolongation of the alkali treatment time up to 60 min resulted in a better improvement in the initial rate of methane production. The initial methane yield has improved by 85% counting up to 0.24 Nm³/kg VS after pretreatment with NaOH for 60 min compared with that of the untreated OPEFB (0.13 Nm³/kg VS).

The accumulated gas production was stabilized after 30 days of incubation and clear positive effects of the pretreatments were

![Fig. 1. FTIR spectra of OPEFB treated with 8% NaOH for 60 min (red) compared with that of the untreated OPEFB (black).](image)

![Fig. 2. Accumulated methane production (Nm³/kg VS) from treated and untreated OPEFB after 7 (grey bars) and 30 days (black bars) of anaerobic digestion.](image)
observed (Fig. 2). 100% improvement in the yield of methane production was achieved when the material was treated with NaOH for 60 min. On the other hand, phosphoric acid treatment resulted in 40% improvement of the methane yield. The highest yield of methane, 0.404Nm³/kg VS, was observed for the OPEFB treated with NaOH for 60 min, counted up to 97% of the theoretical yield of methane on carbohydrates (0.415Nm³/kg carbohydrates) reported by Davidsson (2007).

### 4. Discussion

OPEFB is a byproduct of the main source of vegetable oil production in the world. Only in Malaysia and Indonesia, the annual production of OPEFB is over fifteen and seventeen million tons, respectively. However, today OPEFB is mainly directed into incineration, whereas utilizing this material as a renewable fuel in the transportation sector would give several benefits. A considerable part of the total greenhouse gas emissions originates from fossil transportation fuels (Alavandi and Agrawal, 2008; Sheehan et al., 2003), hence there is a large demand for introducing alternative fuels produced from renewable resources. Moreover, incineration is not a recycling process, since the organic nutrients are not retained and recycled back to soils resulting in removal of structural material from the soil. While digested material left after the anaerobic digestion of OPEFB can be utilized as an excellent and sustainable fertilizer for cultivation of new crops. Additionally, combustion is connected with serious problems such as fly ash disposal and super heater corrosion. Biofuel production from lignocellulosic feedstocks is now under development and could emerge in the future as an important technology with benefits in both renewable energy production and environmental protection.

According to the results of the current work, after treatment with NaOH for 60 min 0.404Nm³ CH₄/kg VS can be produced, so that the methane potential of OPEFB is 316Nm³ CH₄/ton, since OPEFB has a VS content of 79%. This gives the possibility for 4.7 and 5.4 milliard cubic meter methane production in Malaysia and Indonesia, respectively. The energy content of methane is 37MJ/Nm³ (ORNL, 2010), which means an energy production of 11.700 MJ/t OPEFB. Therefore, biometane production from 1t OPEFB can replace around 337L fossil fuels in the transportation sector, assuming the energy content of gasoline or diesel around 35MJ/L (ORNL, 2010). However, an efficient yield of methane can be obtained only after a suitable pretreatment of OPEFB.

Alkaline pretreatment with NaOH has significantly improved both the initial rate of methane production and the final methane yield, by up to 85 and 100%, respectively, under subsequent anaerobic digestion of treated OPEFB samples. OPEFB is a lignocellulosic material with a compact structure, wherein lignin physically shields the cellulosic and hemicellulosic parts. A suitable pretreatment method is therefore needed to open up the structure to make cellulose and hemicelluloses accessible for microbial degradation. NaOH hydrolyzes the bond linkages between lignin and cellulose/hemicellulose as well as intra-lignin linkages. α-ether bonds, phenyl glycosidic linkages, acetal linkages and ester bonds can be cleaved by the added alkali. Furthermore, lignin and hemicellulose are partially solubilized (Durot et al., 2003; He et al., 2008), as it is shown from the decreased values of xylan as well as acid insoluble (AIL) and acid soluble (ASL) lignin in Table 1.

Due to the pretreatment procedures in aquatic solutions the TS and VS content of the treated samples are lower compared to that of the dry untreated OPEFB (Table 2). However, it is beneficial, since there are several advantages with operating the anaerobic digester at low-solid conditions (i.e., TS lower than 15%) among others, lower energy input for pumping and mixing, better accessibility for the microorganisms to the substrates, higher productivity. Moreover, as it is shown in Table 2, the ratio of VS in TS decreased in the pretreatment with phosphoric acid, while it increased with NaOH treatment at all conditions. Therefore, treatment with NaOH is more beneficial for pretreatment of OPEFB not only due to the higher methane yields (Fig. 2), but also due to the higher VS/TS ratios, since it can decrease the reactor volume needed. On the other hand, the VS content in samples treated with NaOH highly dependent on the treatment time, and more VS was lost after longer pretreatment times (Table 2).

In addition, these results indicated that the main parts of material lose during the treatment with acid was organic material, while

### Table 2

<table>
<thead>
<tr>
<th>Material</th>
<th>Total solid (TS) (g/kg)</th>
<th>Ratio of VS in TS (g/kg)</th>
<th>Volatile solids (VS) (g/kg)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Untreated</td>
<td>942</td>
<td>841</td>
<td>792</td>
</tr>
<tr>
<td>H₃PO₄ pretreated</td>
<td>165</td>
<td>788</td>
<td>130</td>
</tr>
<tr>
<td>NaOH pretreated for:</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>10 min</td>
<td>198</td>
<td>980</td>
<td>194</td>
</tr>
<tr>
<td>20 min</td>
<td>124</td>
<td>984</td>
<td>122</td>
</tr>
<tr>
<td>40 min</td>
<td>116</td>
<td>974</td>
<td>113</td>
</tr>
<tr>
<td>60 min</td>
<td>102</td>
<td>863</td>
<td>88</td>
</tr>
</tbody>
</table>

Table 3

<table>
<thead>
<tr>
<th>Wavenumber (cm⁻¹)</th>
<th>Functional group</th>
<th>Assignment</th>
<th>Untreated OPEFB</th>
<th>Treated* OPEFB</th>
</tr>
</thead>
<tbody>
<tr>
<td>3175</td>
<td>-OH stretching intramolecular hydrogen bonds</td>
<td>Cellulose II</td>
<td>0.189</td>
<td>0.199</td>
</tr>
<tr>
<td>2900</td>
<td>C-H stretching</td>
<td>Cellulose</td>
<td>0.301</td>
<td>0.119</td>
</tr>
<tr>
<td>1740</td>
<td>CH stretching of acetyl or carboxylic acid</td>
<td>Hemicellulose and lignin</td>
<td>0.121</td>
<td>0.004</td>
</tr>
<tr>
<td>1610</td>
<td>C=O stretching of the aromatic ring</td>
<td>Lignin</td>
<td>0.198</td>
<td>0.086</td>
</tr>
<tr>
<td>1598</td>
<td>C=C stretching of the aromatic ring</td>
<td>Lignin</td>
<td>0.192</td>
<td>0.105</td>
</tr>
<tr>
<td>1510</td>
<td>Asymmetric bending in C-H</td>
<td>Lignin</td>
<td>0.127</td>
<td>0.086</td>
</tr>
<tr>
<td>1465</td>
<td>Asymmetric bending in C-H</td>
<td>Lignin</td>
<td>0.209</td>
<td>0.156</td>
</tr>
<tr>
<td>1420</td>
<td>C-H₃ symmetric bending</td>
<td>Cellulose</td>
<td>0.227</td>
<td>0.178</td>
</tr>
<tr>
<td>1430</td>
<td>C-H₂ bending</td>
<td>Cellulose</td>
<td>0.227</td>
<td>0.168</td>
</tr>
<tr>
<td>1375</td>
<td>C-H bending</td>
<td>Cellulose</td>
<td>0.251</td>
<td>0.170</td>
</tr>
<tr>
<td>1335</td>
<td>-OH (in plane bending)</td>
<td>Cellulose</td>
<td>0.242</td>
<td>0.186</td>
</tr>
<tr>
<td>1315</td>
<td>C-H₂ wagging</td>
<td>Cellulose</td>
<td>0.254</td>
<td>0.178</td>
</tr>
<tr>
<td>1158</td>
<td>C-O-C asymmetric stretching</td>
<td>Cellulose</td>
<td>0.348</td>
<td>0.262</td>
</tr>
<tr>
<td>898</td>
<td>Asym., out of phase ring stretching (cellulose)</td>
<td>Cellulose</td>
<td>0.280</td>
<td>0.246</td>
</tr>
</tbody>
</table>

* After pretreatment with 8% NaOH for 60 min.
during the treatment with NaOH more inorganic material were lost. However, the relative portions of inorganic and organic material changed with prolongation of NaOH treatment, and resulted in more and more organic materials lose. Therefore, there is a tradeoff between organic materials lose and improvement of biogas production in the alkali treatment and very long treatment time may not be suitable for the treatment in spite of the possible increase in digestibility.

Furthermore, as it is shown from the FTIR analysis data in Table 3, there is a breakdown in the structure and a reduction of crystallinity as results of the above mentioned effects during the treatment. All these facts might together be responsible to improve the attachment of the microbial cellulose (Fontes and Gilbert, 2010), which results in a better digestion process after the treatment (Fig. 2).

5. Conclusions

Biogas production, as an alternative for incineration, can be introduced for sustainable energy production from OPEFB. This lignocellulosic waste material can efficiently be converted to methane after alkali pretreatment with 8% NaOH for 60 min. The pretreatment can modify the structure of carbohydrates in OPEFB making it an easily digestible material.

References


